

Cp₂TiCl₂-Catalyzed Pinacol-Type Coupling of Aliphatic Aldehydes by Use of Zinc and Chlorosilane

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Abstract: Bis(cyclopentadienyl)titanium dichloride (Cp₂TiCl₂) exhibits excellent catalytic activity toward the pinacol-type coupling reaction of aliphatic aldehydes with the assistance of zinc powder and chlorosilane. © 1998 Elsevier Science Ltd. All rights reserved.

The reductive coupling reactions of carbonyl compounds with the aid of low-valent metals is an important method for the construction of a vicinally functionalized carbon-framework.¹ Although numerous reagents have been developed to accomplish the transformation, examples of the catalytic use of low-valent metals are limited to a few cases.² Recently, we have revealed a novel catalytic system consisting of a low-valent vanadium complex, zinc powder, and chlorosilane, which effects the catalytic pinacol-type coupling of aliphatic aldehydes.^{2b,h} During the course of our study on this catalytic system of vanadium, we found that bis(cyclopentadienyl)titanium dichloride (Cp₂TiCl₂) acts as a useful catalyst for the reductive coupling of aldehydes and ketones in the presence of zinc and chlorotrimethylsilane (eq 1).³⁻⁵

R-CHO
$$\frac{\text{cat. Cp}_2\text{TiCl}_2, \text{Zn, Me}_3\text{SiCl}}{\text{solvent, r.t.}} \qquad \begin{array}{c} R & R & R \\ \hline 0 & 0 & + & 0 \\ \hline R & R & R \\ \hline 0 & 0 & + & 0$$

Table 1. Cp₂TiCl₂-Catalyzed Pinacol-Type Coupling of Et₂CHCHO^a

Run	Coreductant	Cat., mol%	Additive	Solvent	2, Yield, % ^b	di / meso ^b
1	Zn	3.0	Me ₃ SiCl	DME	80	67 / 33
2	Zn	3.0		DME	0	
3		3.0	Me ₃ SiCl	DME	0	
4	Zn	1.0	Me ₃ SICI	DME	75	65 / 35
5	Mg	3.0	Me ₃ SiCI	DME	0	
6	Mg	3.0	Me ₃ SiCl	THF	10	78 / 22
7	Zn	3.0	Me ₃ SiCl	THF	64	73 / 27
8	Zn	3.0	Me ₃ SiCl	Et ₂ O	27	35 / 65

^a Reaction conditions: 1 (1.0 mmol), Zn (2.0 mmol), Me₃SiCl (2.0 mmol), solvent (7 mL), r.t., 13 h.

As can be seen in Table 1, the catalytic pinacol-type coupling requires the presence of both zinc powder and chlorosilane⁶ (runs 1-3). The use of magnesium in place of zinc sharply decreased the yield of the coupling product 2 (runs 5-6). The amount of the catalyst could be reduced to 1.0 mol% (run 4). The reductive coupling proceeded efficiently by using DME or THF as a solvent, but not in Et₂O (runs 1, 7-8).

^b Determined by ¹H NMR.

Table 2 represents the results of the Cp_2TiCl_2 -catalyzed reductive coupling of aldehydes and ketones. The procedure can be applied to primary and secondary aldehydes, providing the corresponding oxolanes 2 with moderate and excellent diastereoselectivities, respectively (runs 1-4). In the case of α -tetralone, the initial product, vicinal diol, underwent dehydration to give the corresponding conjugate diene (run 5).

Run	Substrate	Product	Yield, % ^b	di / meso l
1	ⁿ C ₅ H ₁₁ CHO	$2 (R = {}^{n}C_{5}H_{11})$	60	63 / 37
2	PhCH₂CH₂CHO O	2 (R = PhCH ₂ CH ₂)	88	67 / 33
3 ^c	O H	2 (R =	87 (85) ^d	96 / 4
4 ^c	O H	2 (R = -)	66 (53) ^d	81 / 19
5			65 (60) ^d	

Table 2, Cp2TiCl2-Catalyzed Pinacol-Type Coupling®

Next, we demonstrated the intramolecular cyclization of aldehyde 3 bearing a carbon-carbon double bond at an appropriate position by using the cat. Cp_2TiCl_2 / Zn / Me₃SiCl system. The desired reaction proceeded successfully, affording the cyclopentanol derivative 4 in excellent yield, as indicated in eq 2. This cyclization also required the combination of the catalytic system. The absence of Cp_2TiCl_2 or Zn resulted in the complex mixture with the lower selectivity, suggesting that the present cyclization proceeds via a 5-exo cyclization of a ketyl-type intermediate and subsequent chlorine-atom transfer.

In conclusion, we have disclosed a novel catalytic system of Cp₂TiCl₂, which is highly effective for the pinacol-type coupling of aliphatic aldehydes. Further investigation along this line is now in progress.

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References and Notes

- 1. Grame, M. R. In Comprehensive Organic Synthesis; Trost, B. M.; Fleming, I., Ed.; Pergamon: Oxford, 1991; Vol. 3, p 563.
- (a) Fürstner, A.; Hupperts, A. J. Am. Chem. Soc. 1995, 117, 4468. (b) Hirao, T.; Hasegawa, T.; Muguruma, Y.; Ikeda, I. J. Org. Chem. 1996, 61, 366. (c) Nomura, R.; Matsuno, T.; Endo, T. J. Am. Chem. Soc. 1996, 118, 11666. (d) Lipski, T. A.; Hilfiker, M. A.; Nelson, S. G. J. Org. Chem. 1997, 62, 4566. (e) Gansäuer, A. J. Chem. Soc., Chem. Commun. 1997, 457. (f) Gansäuer, A. Synlett 1997, 363. (g) Maury, O.; Villiers, C.; Ephritikhine, M. New. J. Chem. 1997, 21, 137. (h) Hirao, T.; Asahara, M.; Muguruma, Y.; Ogawa, A. J. Org. Chem. 1998, 63, 2812.
- 3. This work was first presented at the 69th Annual Meeting of Chemical Society of Japan, 1995 (March 29).
- 4. For the pinacol coupling reaction of aromatic aldehydes using cat. EBTHITiCl, / Zn / MgBr, / Me₃SiCl system, see ref 2f.
- 5. Zhang Y.; Liu T. Synth. Commun. 1988, 18, 2173.
- Conceivably, Me₃SiCl contributes to both activation of the catalyst and substrates and regeneration of the catalyst.

^a Reaction conditions: 1 (1.0 mmol), Zn (2.0 mmol); activated by the treatment with HCl (1.5 M) and washed with diehtyl ether, purchased from Wako Pure Chemical Industries, Ltd.), Me₃SiCl (2.0 mmol), Cp₂TiCl₂ (3 mol%), DME (7 mL), r.t.,13 h unless otherwise stated. ^b Determined by ¹H NMR. ^c Reaction temperature, 0 °C. ^d Isolated yield.